



Vitamin C Assay Kit

(Cat/No.:BC106 Size 50T/48S)

1. Composition and preparation (Kit Shelf life: 6 months)

Reagent 1: Stock solution, 5 mL × 1 bottle, store at 4 °C. **Preparation of Reagent 1 working solution:** dilute the **Reagent 1 stock solution with distilled water** at a ratio of 1:14 before use and mix well.

Reagent 2: Powder × 1 vial, store at 4 °C. **Preparation of Reagent 2 working solution:** dissolve 1 vial of powder in 0.36 mL glacial acetic acid, then add water to a final volume of 40 mL, dissolve completely, store at 4 °C.

Reagent 3: White crystalline powder × 1 vial, store at 4 °C. **Preparation of Reagent 3 working solution:** due to poor solubility, add 60 mL of 95% or absolute ethanol 2–4 hours before use to dissolve thoroughly, store at 4 °C.

Reagent 4: Yellow stock solution, 0.2 mL × 1 bottle, store at 4–8 °C protected from light. Before use, take 0.15 mL of **Reagent 4 stock solution** and add distilled water to 25 mL to prepare the **Reagent 4 working solution** (or prepare proportionally as needed).

Reagent 5: Liquid, 6 mL × 1 bottle, store at 4 °C.

Reagent 6: Vc standard: powder, 6 mg × 3 vials, each vial contains vitamin C, store at 4 °C protected from light. Before use, dissolve 1 vial of standard powder in 10 mL of **Reagent 1 working solution** to prepare a **600 µg/mL Vc standard working solution**, which can be stored at 4 °C for 1–2 weeks; then dilute the 600 µg/mL Vc standard working solution 100-fold with Reagent 1 working solution to prepare a **6µg/mL Vc standard working solution** for use. Note: Vitamin C standards are easily oxidized; therefore, it is best to prepare them after the sample supernatant has been prepared and allow them to react with reagents within 10 minutes after preparation.

2. Required instruments and reagents

Visible spectrophotometer and cuvettes (or microplate reader (536 ± 10 nm) and 96-well plate), centrifuge, vortex mixer, 37 °C water bath or air incubator, distilled water, absolute ethanol (analytical grade), a small amount of glacial acetic acid (0.5 mL is sufficient, analytical grade), protein assay reagents (for animal tissue samples).

3. Operation procedure

1. Sample pretreatment:

- ① **Animal and plant tissue samples:** Accurately weigh the tissue sample, add 9 volumes of homogenization medium according to the ratio of weight (g): volume (mL) = 1:9 (recommended 0.1 mol/L phosphate buffer with pH 7.0–7.4 or 0.9% physiological



saline), mechanically homogenize in an ice-water bath, centrifuge at 2500 rpm for 10 minutes, and collect the supernatant (10% homogenate supernatant) for testing. **(Note: For animal tissue homogenate supernatant, the protein concentration must be determined (not required for plant tissues; use the second tissue calculation formula during calculation).)**

②. **Serum (plasma) and other liquid samples:** Use directly.

2. Preparation of supernatant: Take 0.15 mL of the above samples (or tissue homogenate supernatant) and add 0.45 mL of **Reagent 1 working solution**, vortex mix, incubate at room temperature for 15 minutes, centrifuge at 3500–4000 rpm for 10 minutes, and collect the clear upper liquid as the supernatant. **(Note: Tissue samples may also be directly homogenized using Reagent 1 working solution as the homogenization medium to prepare supernatant for direct use; refer to the second tissue calculation formula during calculation.)**

3. Determination of Vc in the supernatant:

	Blank tube	Standard tube	Test tube
Reagent 1 working solution (mL)	0.4		
6 µg/mL Vc standard working solution (mL)		0.4	
Supernatant (mL)			0.4
Reagent 2 working solution (mL)	0.5	0.5	0.5
Reagent 3 working solution (mL)	1	1	1
Reagent 4 working solution (mL)	0.25	0.25	0.25
Mix thoroughly, incubate in a 37 °C water bath for 30 minutes			
Reagent 5 (mL)	0.1	0.1	0.1
Mix thoroughly, incubate for 10 minutes, measure absorbance at a wavelength of 536 nm with an optical path length of 1 cm (or 0.5 cm), zero with distilled water, and measure the absorbance of each tube (or take 200 µL of reaction solution from each tube, add to a 96-well plate, and read at 536 nm using a microplate reader).			

4. Calculation formula

1. Calculation and examples for liquid samples:

①. **Calculation formula:**

$$\text{VC content } (\mu\text{g/mL}) = \frac{A_{\text{test}} - A_{\text{blank}}}{A_{\text{standard}} - A_{\text{blank}}} \times C_{\text{standard}} \times N$$

C_{sample} : standard concentration, 6 g/mL;

N : dilution factor before sample measurement (including the pretreatment dilution factor of 4).

②. **Calculation example:**

Example 1: Take 0.15 mL serum and operate according to the operation table. The measured OD values were: blank tube OD = 0.031, standard tube OD = 0.184,



test tube OD = 0.144. The calculation result is:

$$\text{VC content} \left(\frac{\mu\text{g}}{\text{mL}} \right) = \frac{0.144 - 0.031}{0.184 - 0.031} \times 6 \times 4 = 17.73 \mu\text{g}/\text{mL}$$

2. Calculation and examples for tissues:

①. Calculation formula:

$$\begin{aligned} \text{Tissue VC content} \left(\frac{\mu\text{g}}{\text{mgprot}} \right) &= \frac{A_{\text{test}} - A_{\text{blank}}}{A_{\text{standard}} - A_{\text{blank}}} \times C_{\text{standard}} \times N \div \text{Cpr} \\ \text{Tissue VC content} \left(\frac{\mu\text{g}}{\text{g Tissue}} \right) &= \frac{A_{\text{test}} - A_{\text{blank}}}{A_{\text{standard}} - A_{\text{blank}}} \times C_{\text{standard}} \times N \div \frac{W}{V_{\text{sample total}}} \end{aligned}$$

C_{standard}: standard concentration, 6 g/mL;

N: dilution factor before sample measurement (the first formula also includes the pretreatment dilution factor of 4 when Reagent 1 working solution is added);

Cpr: protein concentration of tissue homogenate, mgprot/mL (prot refers to protein);

W: sample weight, g;

V_{sample total}: total volume of homogenization medium added during sample homogenization, mL.

②. Calculation examples:

Example 1: Take 0.15 mL of 10% rat liver homogenate and operate according to the operation table. The measured OD values were: blank tube OD = 0.032, standard tube OD = 0.184, test tube OD = 0.146. The protein concentration of the 10% rat liver homogenate was measured as 8.5 mgprot/mL. The calculation result is:

$$\begin{aligned} \text{VC content in Rat liver} \left(\frac{\mu\text{g}}{\text{mgprot}} \right) &= \frac{0.146 - 0.032}{0.184 - 0.032} \times 6 \times 4 \div 8.5 \\ &= 2.12 \mu\text{g}/\text{mgprot} \end{aligned}$$

Example 2: Take 0.15 mL of 20% rotifer homogenate and operate according to the operation table. The measured OD values were: blank tube OD = 0.031, standard tube OD = 0.186, test tube OD = 0.177. The protein concentration of the 20% rotifer homogenate was measured as 0.854 mgprot/mL. The calculation result is:

$$\begin{aligned} \text{Rotifer VC content} \left(\frac{\mu\text{g}}{\text{mgprot}} \right) &= \frac{0.177 - 0.031}{0.186 - 0.031} \times 6 \times 4 \div 0.854 \\ &= 26.48 \mu\text{g}/\text{mgprot} \end{aligned}$$

5. Assay principle

In this method, Fe³⁺ rapidly reacts with reduced ascorbic acid to generate Fe²⁺, which subsequently reacts with o-phenanthroline to produce a colored complex, allowing the determination of vitamin C content in plasma.

6. Assay significance

Vitamin C participates in collagen synthesis in organisms and is a cofactor for proline hydroxylase and lysine hydroxylase. It is also required for the copper-containing enzyme dopamine β-hydroxylase, which catalyzes the conversion of dopamine to norepinephrine. When vitamin C is deficient, synthesized collagen cannot be sufficiently hydroxylated and cannot



properly form fibers. The most prominent chemical activity of vitamin C is its role as a reducing agent, whereby it reduces Fe^{3+} to Fe^{2+} , promoting intestinal iron absorption and enhancing iron storage and utilization. As a free radical scavenger, ascorbic acid can rapidly react with O_2^- and H_2O_2 , and even more rapidly with OH^- to form ascorbyl radicals. It can also scavenge O_2^- , thereby protecting the organism from damage caused by endogenous oxygen free radicals. Vitamin C can reduce α -tocopheroxyl radicals back to vitamin E, indirectly acting as a chain-breaking antioxidant. Vitamin C is the most effective antioxidant in plasma and constitutes the first line of defense of the extracellular antioxidant defense system. It also inhibits ongoing lipid peroxidation in plasma.